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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO	
10/849,721	05/20/2004	Mark A. Hochwalt	713629.421	8654	
27128 7590 03/29/2005 BLACKWELL SANDERS PEPER MARTIN LLP			EXAMINER CHOI, FRANK I		
ST. LOUIS, MO 63101			. 1616		
			DATE MAILED: 03/29/2005		

Please find below and/or attached an Office communication concerning this application or proceeding.

		Applicati	on No.	Applicant(s)				
Office Action Summary		10/849,7	21	HOCHWALT ET AL.				
		Examine		Art Unit				
		Frank I. C	****	1616				
The MA Period for Reply	AILING DATE of this communic	cation appears on th	ecover sheet with the co	orrespondence address	S			
THE MAILING - Extensions of time after SIX (6) MON - If the period for re If NO period for re Failure to reply with Any reply received	ED STATUTORY PERIOD FO DATE OF THIS COMMUNION e may be available under the provisions of ITHS from the mailing date of this commu- ply specified above is less than thirty (30 eply is specified above, the maximum state thin the set or extended period for reply of d by the Office later than three months after an adjustment. See 37 CFR 1.704(b).	CATION. of 37 CFR 1.136(a). In no evunication. ) days, a reply within the state tutory period will apply and will, by statute, cause the app	ent, however, may a reply be time utory minimum of thirty (30) days ill expire SIX (6) MONTHS from t lication to become ABANDONEC	ety filed will be considered timely. the mailing date of this commun (35 U.S.C. § 133).	nication.			
Status								
1)⊠ Respons	sive to communication(s) filed	d on						
2a)☐ This acti	ion is <b>FINAL</b> . 2	b)⊠ This action is r	on-final.	•				
•	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.							
Disposition of Cla	aims							
4a) Of th 5)	Claim(s) <u>1-49</u> is/are pending in the application.  4a) Of the above claim(s) is/are withdrawn from consideration.  Claim(s) is/are allowed.  Claim(s) <u>1-49</u> is/are rejected.							
Application Pape	rs							
9)☐ The spec	cification is objected to by the	Examiner.						
	☐ The drawing(s) filed on is/are: a)☐ accepted or b)☐ objected to by the Examiner.							
• •	Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).							
·	Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).  1) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Priority under 35	U.S.C. § 119							
a)	edgment is made of a claim f ) Some * c) None of: ertified copies of the priority of pries of the certified copies of oplication from the Internation ttached detailed Office action	documents have bee documents have bee of the priority docum nal Bureau (PCT Ru	en received. en received in Application ents have been receive e 17.2(a)).	on No  In this National Stag	j <b>e</b>			
2) Notice of Draftsp 3) Information Disc	nces Cited (PTO-892) person's Patent Drawing Review (PT		• —		)			
Paper No(s)/Mai	ı Dale		6)					

Application/Control Number: 10/849,721

Art Unit: 1616

## **DETAILED ACTION**

## Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-49 are rejected under 35 U.S.C. 103(a) as being unpatentable over Marcus et al. (US Pat. 4,826,497) in view of EP 0 509 409, Peterson et al. (US Pat. 5,780,020), JP 63216572 (Abstract), JP 72046908 (Abstract) and DE 19837539 (Abstract).

Marcus et al. teach fibrous absorbent articles containing a zeolite having at least about 90 percent framework of tetrahedral oxides units being Si02 tetrahedra, a sorptive capacity for water of less than 10 weigh percent when measure at 25 degrees Celsius and 4.6 torr, and a pore diameter of at least 5.5 Angstroms, where the water of hydration is removed (Column 3, lines 40-68, Column 4, lines 1-35). It is taught that other components such as medicants, other absorbents and adsorbents, such as sodium bicarbonate and clinoptilolite (Column 7, lines 23-29). It is taught that clinoptilolite is a preferred odor suppressant (Column 3, lines 32-35). It is disclosed that high zeolites, such as Abscents®, are preferred over the intermediate zeolites for control of odors associated with sulfur compounds (Column 17, lines 60-68, Column 18, lines 1-5).

EP 0 509 409 teach feminine products having a combination of acidic, basic and neutral deodorants and that that acidic particles having a pH of less than 7 are suitable as deodorants of basic compounds, that sodium bicarbonate is suitable as a deodorant of acidic compounds and

that Abscents® and Smellrite ® are preferred pH neutral odor absorbing materials as they retain most of their absorbent capacity in moist conditions (pgs. 4,5).

Peterson et al. teach that antimicrobial agents, such as zinc oxide, are used in deodorant products to help to control odor by controlling bacteria or fungi (Column 4, lines 50-68, Column 5, lines 1-23). Peterson et al. discloses that the zeolites of the claimed invention are suitable for use in invention of Peterson et al. (See Peterson et al., Column 4, lines 58-61).

JP 63216572 (Abstract) teaches that sorbic acid is an effective deodorant.

JP 72046908 (Abstract) teaches that itaconic acid is an effective deodorant.

DE 19837539 (Abstract) teaches that maleic or fumaric acids are effective deodorants.

The difference between the prior art and the claimed invention is that the prior art does not expressly disclose the combination of the claimed acids, the claimed synthetic zeolite and metal or metal oxides. However, the prior art amply suggests the same as it is known the deodorant products contain the claimed acids, the claimed synthetic zeolites and the claimed metal or metal oxides for odor control. As such, it would have been well within the skill of and one of ordinary skill in the art would have been motivated to modify the prior art as above with the expectation that the addition of the acids, sodium bicarbonate, clinoptilolite and metal oxides would improve odor control.

One of ordinary skill in the art would be motivated to add zinc oxide as zinc oxide is disclosed to control odors by acting as an antimicrobial agent, would be motivated to add an acid having a pH of less than 7 as said acids are disclosed to control odors from basic compounds and would be motivated to use Smellrite® or Abscent® as the same are disclosed to retain most of

Art Unit: 1616

their absorbent capacity in moist conditions with the expectation that the combination would be more effective as the combination would work against odors resulting from various sources.

The claimed acids are clearly encompassed by the disclosure of EP 0 509 409. EP 0 509 409 specifically discloses the following acids, maleic acid, stearic acid and malonic acid. Maleic acid appears to be structurally similar to aspartic acid (adds amino group substituent to carbon chain, removes double bond), methylsuccinic acid (adds methyl group substitutent to carbon chain, removes double bond), adipic acid (lengthens carbon chain by two methyl groups, removes double bond), glutaric acid (lengthens carbon chain by one methyl group, removes double bond), itaconic acid (adds double bond methyl group substituent, removes double bond), tartaric acid (adds two hydroxyl group substituents, removes double bond), and fumaric acid (trans isomer). Stearic acid appears structurally similar to sorbic acid (reduces carbon chain by 12 carbons, adds two double bonds). Malonic acid appears structurally similar to dimethymalonic acid (adds two methyl group substituents to carbon chain), adipic acid (increases carbon chain by 2 carbons), glutaric acid (increases carbon chain by 1 carbon) and fumaric acid (increases carbon chain by 1 carbon, adds double bond). As indicated above, the acids disclosed in EP 509 409 are disclosed to be effective in controlling odors. Further, as indicated above, the specifically disclosed acids in EP 509 409 appear to be structurally similar to the claimed acids and the prior art teaches that sorbic acid, itaconic acid and fumaric acid are effective deodorants. "An obviousness rejection based on similarity in chemical structure and function entails the motivation of one skilled in the art to make a claimed compound, in the expectation that compounds similar in structure will have similar properties." In re Payne, 203 USPQ 245, 254 (CCPA 1979). See also In re Dillon, 16 USPQ2d 1897 (Fed. Cir. 1991). As

Application/Control Number: 10/849,721 Page 5

Art Unit: 1616

such, one of ordinary skill in the art would expect that homologues of maleic acid, stearic acid and malonic acid, including adipic, aspartic, dimethylmalonic, fumaric, sorbic, glutaric, methylsuccinic, itaconic or tartaric acid, would have similar deodorant activity.

"It is prima facie obvious to combine two compositions each of which is taught by the prior art to be useful for the same purpose, in order to form a third composition to be used for the very same purpose.... [T]he idea of combining them flows logically from their having been individually taught in the prior art." In re Kerkhoven, 205 USPQ 1069, 1072 (CCPA 1980) (citations omitted) (Claims to a process of preparing a spray-dried detergent by mixing together two conventional spray-dried detergents were held to be prima facie obvious.). See also In re Crockett, 126 USPQ 186 (CCPA 1960) (Claims directed to a method and material for treating cast iron using a mixture comprising calcium carbide and magnesium oxide were held unpatentable over prior art disclosures that the aforementioned components individually promote the formation of a nodular structure in cast iron.); and Ex parte Quadranti, 25 USPQ2d 1071 (Bd. Pat. App. & Inter. 1992) (mixture of two known herbicides held prima facie obvious). As such, it would be obvious to combine deodorants to together with the expectation that the combination would also have deodorant properties.

Therefore, the claimed invention, as a whole, would have been *prima facie* obvious to one of ordinary skill in the art at the time the invention was made, because every element of the invention has been collectively taught by the combined teachings of the references.

## Conclusion

A facsimile center has been established in Technology Center 1600. The hours of operation are Monday through Friday, 8:45 AM to 4:45 PM. The telecopier number for accessing the facsimile machine is 571-273-8300.

Page 6

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Frank Choi whose telephone number is (571)272-0610. Examiner maintains a flexible schedule. However, Examiner may generally be reached Monday-Friday, 8:00 am - 5:30 pm (EST), except the first Friday of the each biweek which is Examiner's normally scheduled day off.

If attempts to reach the Examiner by telephone are unsuccessful, the Examiner's Supervisor, Mr. Gary Kunz, can be reached at 571-272-0887. Additionally, Technology Center 1600's Receptionist and Customer Service can be reached at (571) 272-1600.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). FIC March 21, 2005

S. MARK CLARDY PATENT EXAMINER GROUP 1209

1616